

Course Outline

Climate 101

- 2/7 Introduction: The Earth System
- **2/14 Energy, Radiation, & Temperature**
- 2/21 Winds, Currents, and Water
- 2/28 Climates of the Past
- 3/7 Modern Climate Change

Class Web Site

<http://climate101.atmos.colostate.edu>

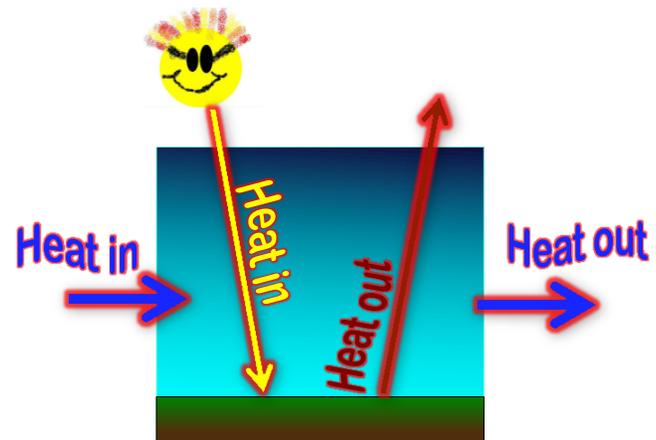
- All slides as printable handouts
- Supplemental readings
- Videos
- Links to more resources

Weather vs Climate

what's the difference?

- If you don't like the **weather**:
 - *Wait five minutes!*
- If you don't like the **climate**:
 - *Move!*

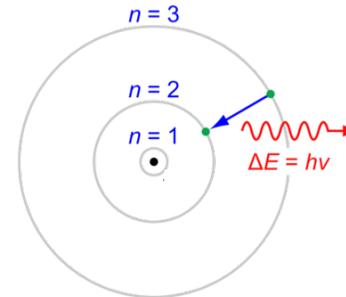
Heat Budgets



Radiation and the Planetary Energy Balance

- Electromagnetic Radiation
- Solar radiation warms the planet
- Conversion of solar energy at the surface
- Absorption and emission by the atmosphere
- The greenhouse effect
- Planetary energy balance

Waves and Photons



Long Waves = small photons
Short Waves = BIG PHOTONS

Is light a wave?
YES!

Is light a particle?
YES!

All light travels at the same speed

Think of short waves as BIG HEAVY particles

Think of longer waves as small, lightweight particles

Defining Energy is Hard!

- “**Energy** is the capacity to perform **work**”
 - (but physicists have a special definition for “work,” too!)
- Part of the trouble is that scientists have “appropriated” common English words and given them **special meanings**
- But part of the trouble is that the concept is **absolutely central to understanding the physical world**, so deeply buried in our language it’s hard to get our heads around

Conservation of Energy

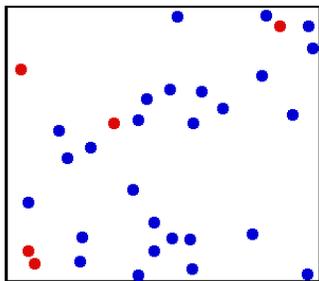
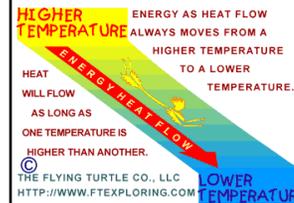
- Energy can be **stored**
- Energy can **move** from one piece of matter to another piece of matter
- Energy can be **transformed** from one type of energy to another type of energy
- **The First Law of Thermodynamics:**
 - During all this moving and transforming the total amount of energy never changes.

Kinds of Energy

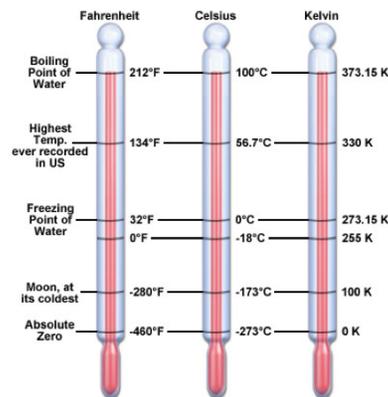
- Radiant Energy -- light
- Kinetic Energy -- motion
- Gravitational Potential Energy -- height
- “Internal Energy”
 - Temperature, Pressure -- hot air
 - Chemical energy
 - Nuclear energy
- **Transfers and conversions among different kinds of energy power everything that happens in our climate!**

If Energy is Conserved ... then why do we need to “conserve energy?”

- Total energy is conserved (First Law), but not its **usefulness!**
- **Second Law of Thermodynamics:**
Energy flows “downhill” from highly concentrated (hot) forms to very dilute (cold) forms
- Gasoline burned in your car (hot) makes it move
- Turbulence and friction of tires on road dissipated as heat
- Heat radiated to space (cold)



Temperature is Motion

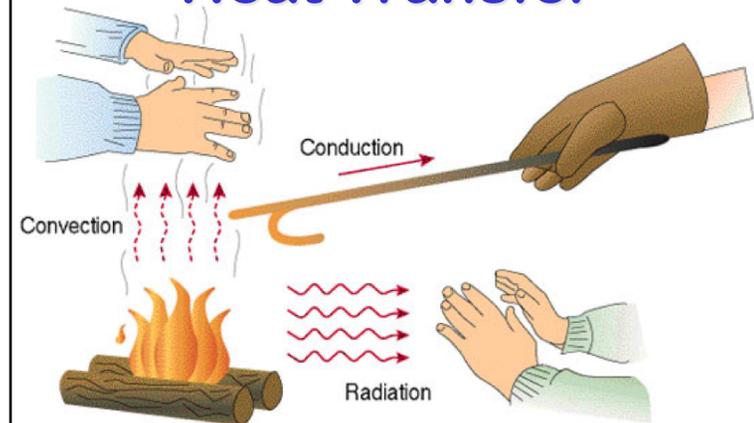


Temperature is a measure of the **average kinetic (motion) energy** of molecules

- $K.E. = \frac{1}{2} mv^2$
m = mass, v = velocity
- So...**temperature is a measure of air molecule speed** (squared)

Thermometers are speedometers!

Heat Transfer



Conduction is by hot molecules colliding with neighbors
Convection is by hot stuff moving in bulk from place to place

Electromagnetic Radiation

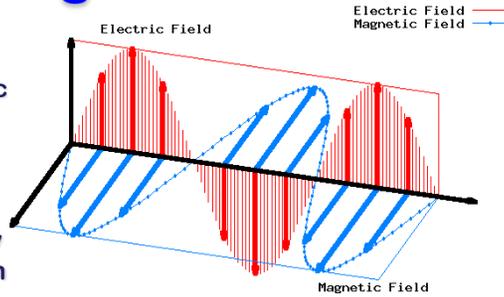
Changing electric fields create changing magnetic fields ...

and vice versa!

This makes energy move even through empty space

We can see it, feel it

Plants harvest it directly, and we harvest them!



Travels at 3×10^8 m/s
= 186,000 miles / sec !

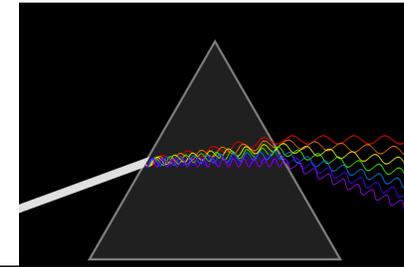
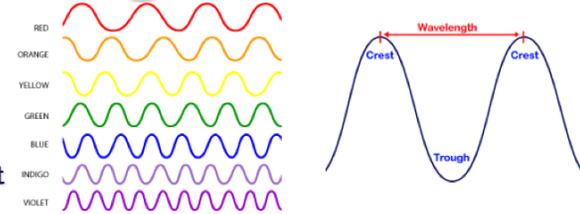
Distance it goes in one cycle is called the wavelength

Electromagnetic Radiation

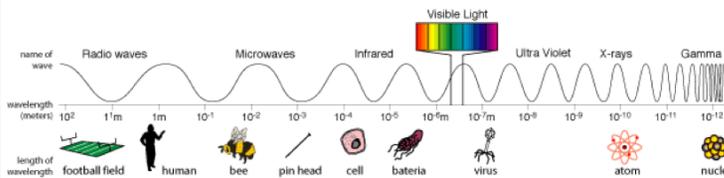
Radiation travels as waves or photons

Waves do not require molecules to propagate

Shorter waves carry more energy than longer ones



Electromagnetic Radiation Spectrum



Shorter waves carry more energy than longer waves

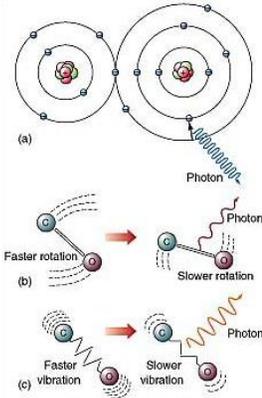
Electromagnetic waves interact with matter at similar scales (sizes) as the waves

Thermal Radiation



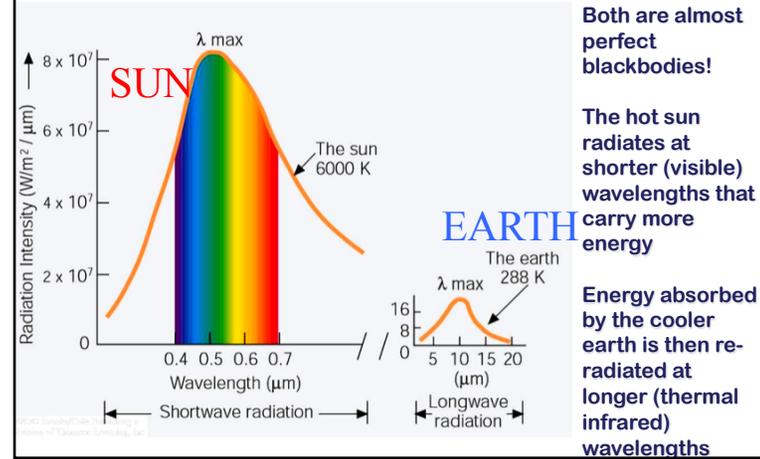
"Everything emits heat"

Atoms, Molecules, and Photons



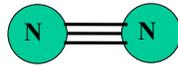
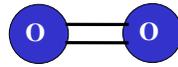
- Atmospheric gases are made of **molecules**
- Molecules are groups of atoms that **share electrons (bonds)**
- **Photons** can interact with bonds
- Transitions between one state and another involve **specific amounts of energy**

Thermal Emission



Dancing Molecules & Heat Rays!

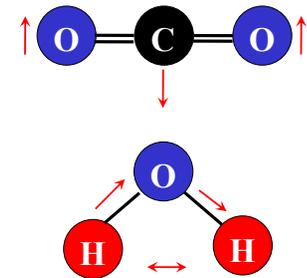
- Nearly all of the air is made of oxygen (O_2) and nitrogen (N_2) in which **two atoms of the same element** share electrons
- Infrared (heat) **energy radiated up from the surface can be absorbed** by these molecules, but not very well



Diatomic molecules can vibrate back and forth like balls on a spring, but the ends are identical

Dancing Molecules & Heat Rays!

- Carbon dioxide (CO_2) and water vapor (H_2O) are different!
- They have **many more ways to vibrate** and rotate, so they are very good at absorbing and emitting infrared (heat) radiation



Molecules that have many ways to wiggle are called "Greenhouse" molecules

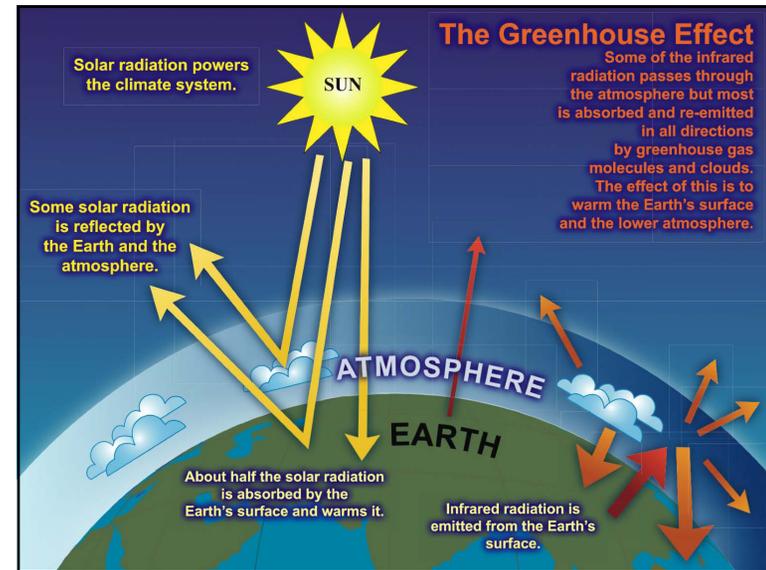
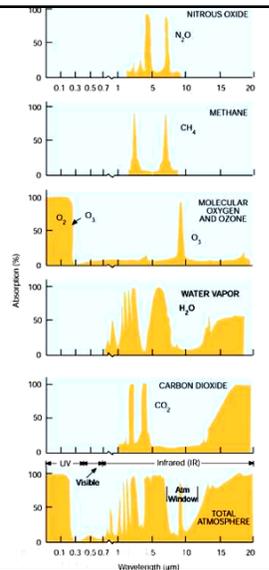
Atmospheric Absorption

Solar radiation passes rather freely through Earth's atmosphere

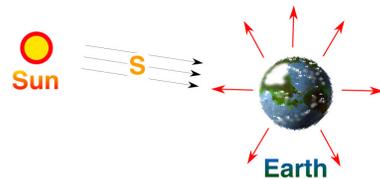
Earth's re-emitted longwave energy either fits through a narrow "window" or is absorbed by greenhouse gases and re-radiated toward earth

Major LW absorbers:

- Water vapor
- CO₂
- O₃
- Clouds



Planetary Energy Balance



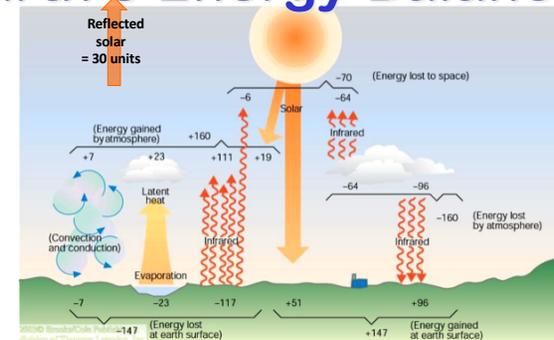
Energy In = Energy Out

$$S(1-\alpha)\pi R^2 = 4\pi R^2\sigma T^4$$

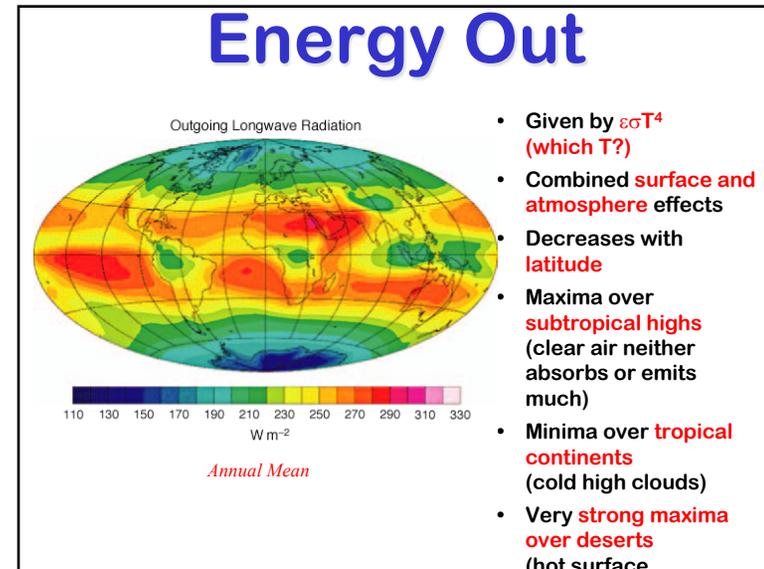
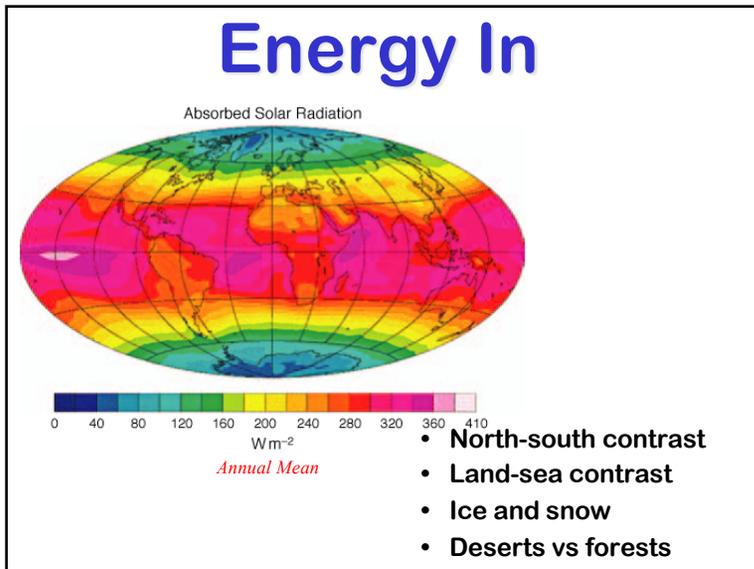
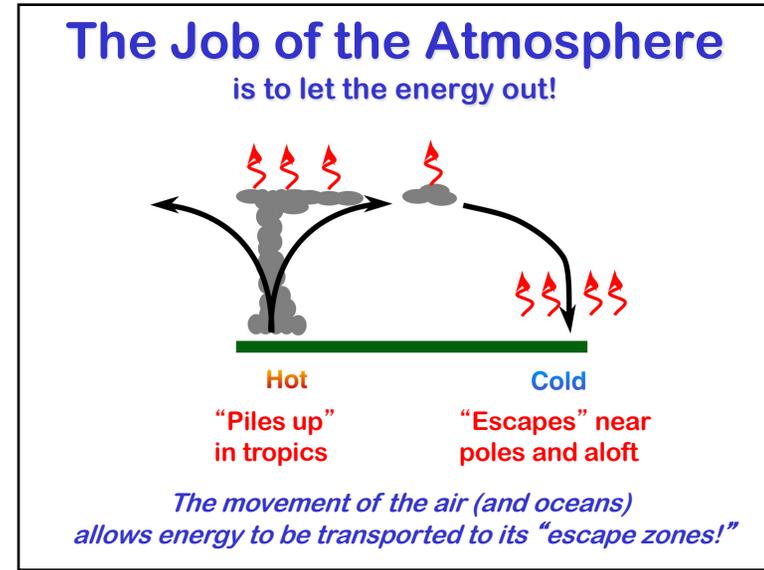
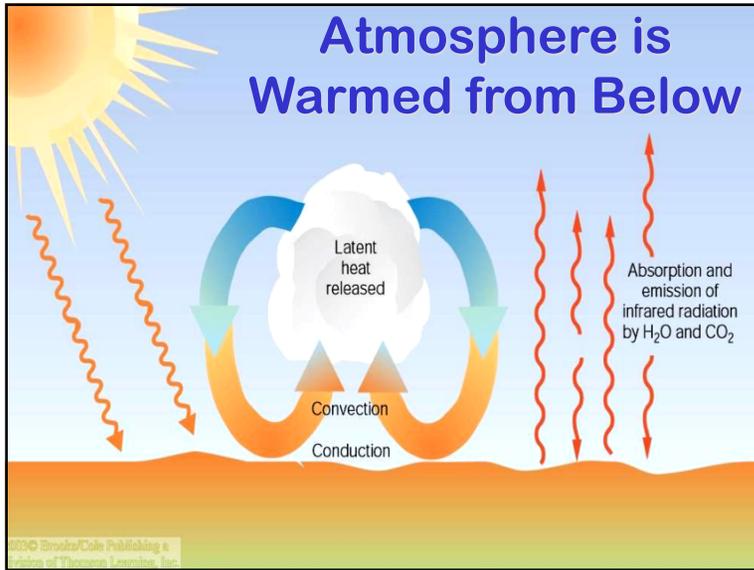
$$T \approx -18^\circ\text{C}$$

But the observed T_s is about 15°C

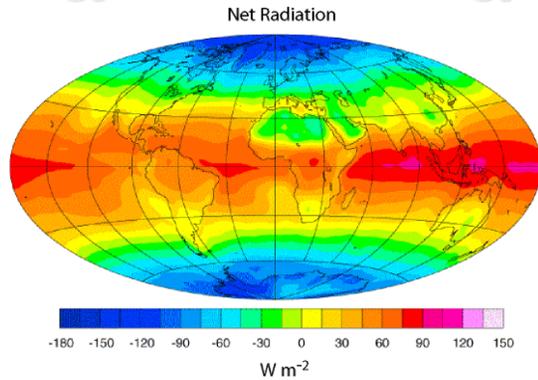
Earth's Energy Balance(s)



- Surface absorbs **51 units of sunshine**, plus **96 units of thermal IR!** (total = 147 units, **47% more than incoming solar!**)
- Surface emits only 117 units, gives the rest back by evaporating water (23 units) and convection (7 units)



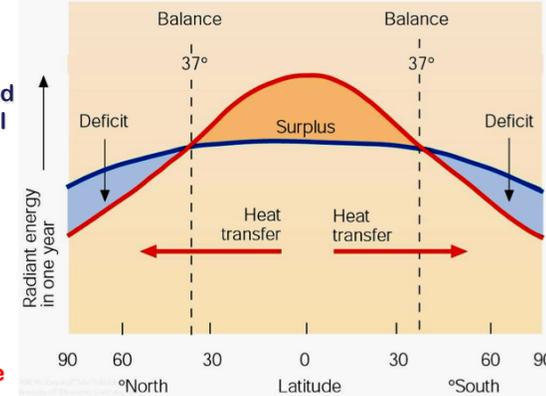
Energy In minus Energy Out



- Incoming solar minus outgoing longwave
- Must be balanced by horizontal transport of energy by atmosphere and oceans!

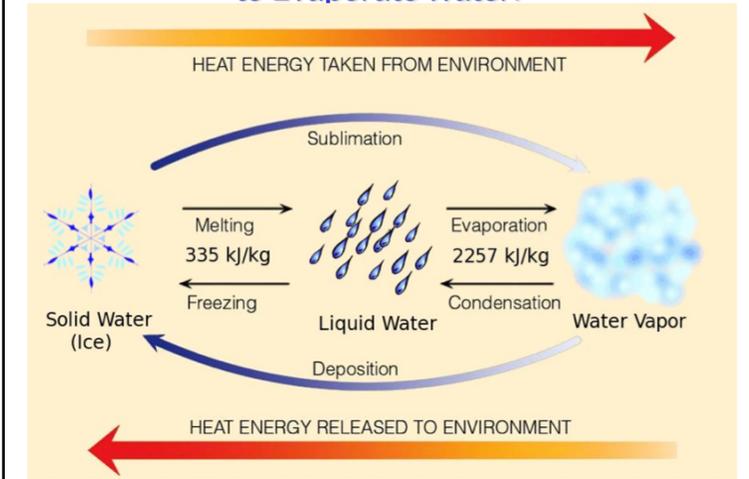
Earth's Energy Balance

Earth's annual energy balance between solar insolation and terrestrial infrared radiation is global but not local

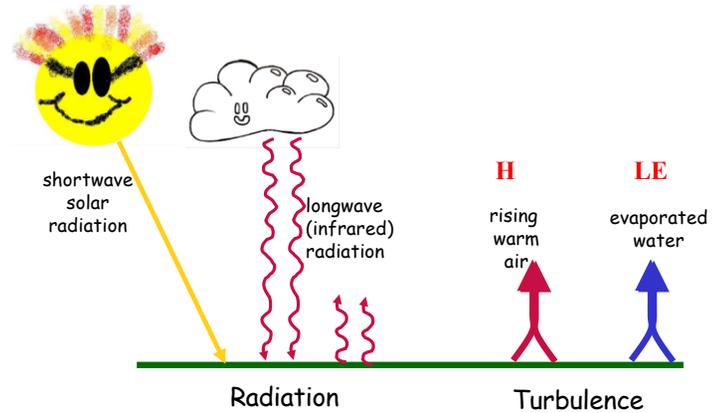


The global balance is maintained by transferring excess heat from the equatorial region toward the poles

It Takes a Lot of Energy to Evaporate Water!

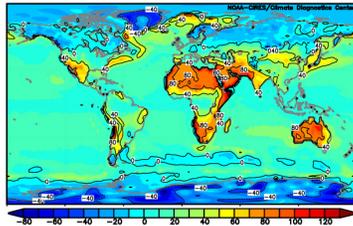


Energy Balance of Earth's Surface

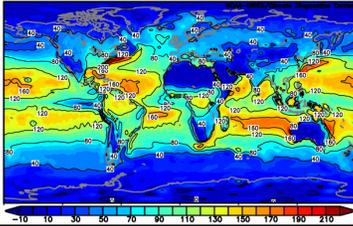


Energy from the Surface to the Air

Rising Warm Air (H)

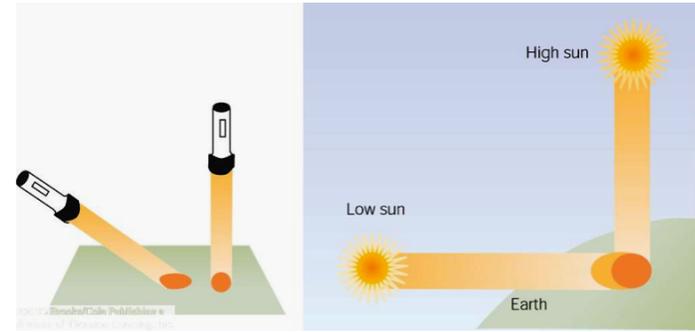


Evaporated Water (LE)



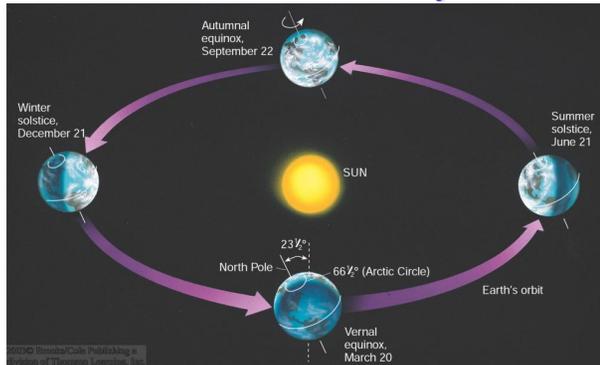
- Energy absorbed at the surface warms the air
- Some of this energy is transferred in rising warm “thermals”
- But more of it is “hidden” in water vapor

Seasons & Solar Intensity



A sunlight beam that strikes at an angle is spread across a greater surface area, and is a less intense heat source than a beam impinging directly.

Solstice & Equinox



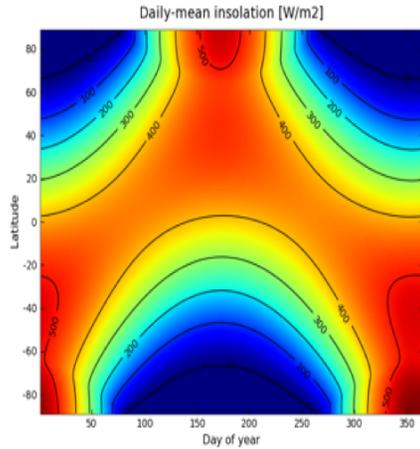
- At **solstice**, one pole is light & one is dark (24/7)
- At **equinox**, tilt provides exactly **12 hours of night and 12 hours of day** everywhere

Midnight Sun



The region north of the Arctic Circle experiences a period of 24 hour sunlight in summer, where the Earth's surface does not rotate out of solar exposure

Daily Solar at Top of Atmosphere



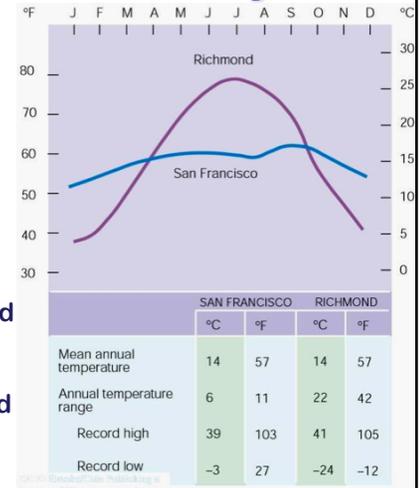
- 75° N in June gets more sun than the Equator!
- Compare N-S changes by seasons
- Very little tropical seasonality

Regional Seasonal Cycles

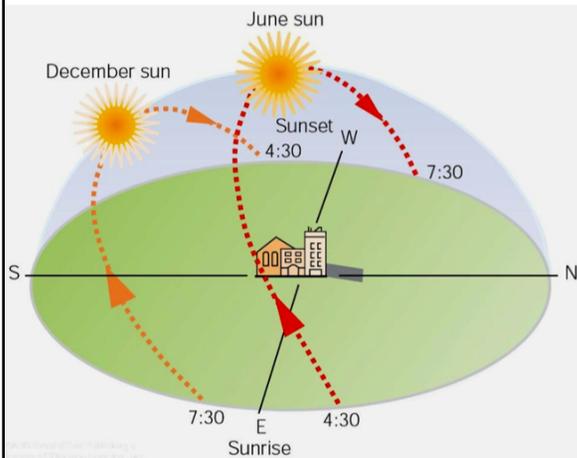
Regional differences in temperature, from annual or daily, are influenced by geography, such as latitude, altitude, and nearby water or ocean currents, as well as heat generated in urban areas

San Francisco is downwind of the Pacific Ocean

Richmond, VA is downwind of North America!



Local Solar Changes



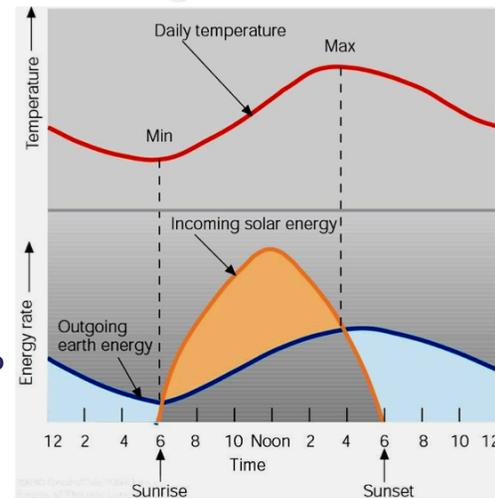
Northern hemisphere sunrises are in the southeast during winter, but in the northeast in summer

Summer noon time sun is also higher above the horizon than the winter sun

Temperature Lags Radiation

Earth's surface temperature is a balance between incoming solar radiation and outgoing terrestrial radiation.

Peak temperature lags after peak insolation because surface continues to warm until infrared radiation exceeds insolation.



Remember

- All energy exchange with Earth is **radiation**
- Outgoing radiation has longer waves (cooler)
- Longwave radiation is **absorbed and re-emitted by molecules** in the air (H₂O & CO₂)
- Atmosphere is selectively transparent! Short waves get through, but long waves get trapped
- Recycling of energy between air and surface is the **“greenhouse effect”**
- Changes of **angle of incoming sunlight** and length of day & night are responsible for seasons and for north-south differences in climate
- Regional energy surpluses and deficits drive the **atmosphere and ocean circulations** (wind & currents)